Katwa College Department of Chemistry

Internal Assessment Examination, 2023

Semester: V Paper: DSE-2 Full Marks: 10

Answer the following questions

2+2+2+2+2

- 1. Calculate the mean and the standard deviation of the following set of analytical results: 15.67, 15.69, and 16.03 g.
- 2. Write down two significance of ε in Beer-Lambert's law.
- 3. 20 mg mixture of CaCO₃ and MgCO₃ are heated up to 600°C and observed 22% of weight loss near 400°C in thermogravimetric analysis. What is the ratio of Ca/Mg in the mixture?
- 4. Prove that multiple extraction is more efficient than a single extraction provided total volume of organic and aqueous phases and same in both the cases.
- 5. Write down the stationary phase and the mobile phase of Paper chromatography and TLC.

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